

## Schoonerite: its atomic arrangement

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## Abstract

Schoonerite,  $\text{ZnMn}^{2+}\text{Fe}_2^{3+}\text{Fe}^{3+}(\text{OH})_2(\text{H}_2\text{O})_7(\text{PO}_4)_3 \cdot 2\text{H}_2\text{O}$ , orthorhombic,  $a$  11.119(4),  $b$  25.546(11),  $c$  6.437(3) Å,  $Pmab$ ,  $Z = 4$ , is one of several structurally-related phosphate minerals having axial repeats approximating 6.4 Å. The structure was solved by Patterson and Fourier syntheses, and least-squares refinement converged to  $R(hkl) = 0.085$  for 897 nonequivalent reflections.

Edge-sharing  $\text{Fe}^{2+}$ -O octahedral chains join by sharing corners with  $\text{Fe}^{3+}$ -O octahedra and  $[\text{PO}_4]$  tetrahedra forming a sheet parallel to  $\{010\}$ . To this sheet are linked additional  $[\text{PO}_4]$  tetrahedra,  $[\text{MnO}_6]$  octahedra and  $[\text{ZnO}_5]$  polyhedra. The resultant slabs are joined in the  $b$  direction by a network of hydrogen bonds only. Interlayer water molecules contribute to this hydrogen bond network.

Average interatomic distances are <sup>6</sup>  $\text{Fe}^{2+}(1)\text{-O}$  2.12 Å, <sup>6</sup>  $\text{Fe}^{2+}(2)\text{-O}$  2.15, <sup>6</sup>  $\text{Fe}^{2+}(3)\text{-O}$  2.00, <sup>6</sup>  $\text{Mn}^{2+}\text{-O}$  2.16, <sup>5</sup>  $\text{Zn-O}$  2.09, <sup>4</sup>  $\text{P}(1)\text{-O}$  1.54, and <sup>4</sup>  $\text{P}(2)\text{-O}$  1.54.

## Introduction

In the preceding paper, Moore and Kampf (1977) described the new species, schoonerite,  $\text{ZnMn}^{2+}\text{Fe}_2^{3+}\text{Fe}^{3+}(\text{OH})_2(\text{H}_2\text{O})_7(\text{PO}_4)_3 \cdot 2\text{H}_2\text{O}$ ,  $Z = 4$ , from the Palermo No. 1 pegmatite in North Groton, New Hampshire. Although this species is rather widely distributed at Palermo, it occurs only sparingly as sprays of small, thin, feathery crystals in pockets and on fracture surfaces. A water determination was deemed unfeasible due to the difficulty in obtaining a sufficient quantity of pure material. The determination of the crystal structure of schoonerite was considered the best method for obtaining an accurate measure of its water content and for characterizing its rather complex crystal chemistry.

Besides accomplishing these ends, the crystal structure determination revealed a close similarity between the structures of schoonerite and montgomeryite, and in addition showed that schoonerite possesses octahedral and tetrahedral clusters locally similar to those in olmsteadite and melonjosephite.

## Experimental

Good single crystals of schoonerite are exceedingly rare, and such crystals large enough for structure

analysis seemed for quite some time nonexistent. The crystal chosen, a thin rectangular tablet measuring  $0.32 \times 0.01 \times 0.08$  mm along the three crystallographic axes, was mounted with the  $a$  axis parallel to the  $\phi$  axis on a Picker automated diffractometer. The intensities of 2383 reflections (maximum  $2\theta = 45^\circ$ ) were gathered, utilizing graphite-monochromatized  $\text{MoK}\alpha$  radiation. A rather large mosaic spread necessitated wide half-angle scans of  $1.0^\circ$  with a scan rate of  $2.0^\circ/\text{minute}$ . Twenty-second background counting times were used on either side of each reflection. The least-squares refinement of 30 reference reflections ( $2\theta = 20\text{--}30^\circ$ ) yielded cell constants,  $a$  11.119(4),  $b$  25.546(11), and  $c$  6.437(3) Å. Systematic extinctions were consistent with either of the space groups  $P2ab$  or  $Pmab$ .

The measured intensities were corrected for absorption using the Gaussian integral method described by Burnham (1966). The data were processed by conventional computational procedures. After averaging symmetry-equivalent reflections and rejecting all  $I(hkl) < 3\sigma [I(hkl)]$ , 897 independent data were available for the ensuing analysis.

## Solution and refinement of the structure

The Patterson map,  $P(uvw)$ , revealed strong vector concentrations at the levels  $u = 0, \frac{1}{4}$ , and  $\frac{1}{2}$ , indicating that the twenty heavy atoms (Fe, Mn, Zn) in

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10	5	930	804	1662	1541	10	6	812	858	542	1035	1012	1060	581	492	511
10	10	760	679	1662	679	10	6	500	444	274	350	888	531	245	252	211
12	1	426	324	1259	1259	10	5	658	618	304	260	362	349	1134	1221	113
12	2	1264	1160	1264	1160	10	6	388	266	538	856	354	245	570	1021	576
12	3	1744	1681	1744	1681	11	6	320	459	265	116	541	475	512	524	416
12	4	1346	1320	1744	1681	11	1	843	525	1767	1810	435	471	347	310	233
12	5	1538	1285	1346	1320	11	2	752	808	554	606	354	230	476	423	574
12	6	479	380	1538	1285	11	3	386	355	1266	1425	496	434	308	400	487
12	7	509	941	479	380	11	4	244	385	1024	1023	446	182	445	700	405
12	8	1203	1150	509	941	11	5	1026	323	3154	1224	255	368	779	718	545
12	9	824	759	1203	1150	11	6	1026	245	341	401	464	260	278	345	438
14	0	3185	460	824	759	11	7	340	239	637	605	258	791	276	230	700
14	1	1345	3125	3185	460	11	8	327	235	401	542	258	286	445	438	405
14	2	1201	1123	1345	3125	12	8	365	400	548	526	344	300	715	750	438
14	3	1526	1525	1201	1123	12	9	251	243	1353	1378	595	256	232	242	438
14	4	1768	1742	1526	1525	12	0	515	417	550	595	502	1234	683	120	438
14	5	875	753	1768	1742	13	0	372	255	401	465	1251	1234	683	120	438
14	6	419	278	875	753	13	1	1084	370	250	140	776	751	445	445	438
14	7	1552	1522	419	278	13	2	703	727	1038	1060	1524	1266	232	242	438
14	8	1269	1151	1552	1522	13	3	434	458	1180	1218	1241	1266	232	242	438
14	9	1639	1523	1269	1151	13	4	548	505	944	978	1137	1180	545	545	438
16	0	1418	1374	1639	1523	13	5	1272	602	1471	1495	830	818	325	354	581
16	1	1510	1484	1418	1374	13	6	433	428	1261	2225	797	563	428	390	438
16	2	833	913	1510	1484	13	7	654	714	455	511	504	505	355	297	547
16	3	644	658	833	913	13	8	350	240	1658	1658	504	505	355	297	547
16	4	1392	1272	644	658	14	8	456	414	765	750	336	408	690	720	425
16	5	1663	1566	1392	1272	14	9	483	362	272	338	437	451	635	677	425
16	6	704	714	1663	1566	14	0	301	277	1565	1585	446	765	437	451	425
16	7	1130	1036	704	714	14	1	282	221	1215	1183	246	303	275	252	275
16	8	553	588	1130	1036	14	2	1047	639	480	525	338	408	690	720	425
16	9	1613	1603	553	588	14	3	551	500	2152	2184	511	1589	1555	1555	425
18	0	545	466	1613	1603	15	0	935	535	530	511	746	739	695	695	425
18	1	470	340	545	466	15	1	657	647	356	316	1056	1105	1167	1213	425
18	2	309	228	470	340	15	2	681	644	782	821	1670	1647	755	788	425
18	3	584	858	309	228	15	3	503	430	747	769	554	647	765	788	425
18	4	975	621	584	858	15	4	726	655	355	335	289	219	615	608	425
20	0	441	406	975	621	15	5	616	615	380	341	289	219	615	608	425
20	1	507	528	441	406	15	6	268	152	1592	1631	513	553	593	611	491
20	2	291	156	507	528	15	7	602	506	895	856	605	578	404	424	491
20	3	564	215	291	156	15	8	585	694	2075	2086	665	650	341	424	491
20	4	535	480	564	215	16	8	518	557	1285	1382	356	235	458	512	491
20	5	835	767	535	480	16	9	688	671	1147	1114	715	683	685	685	491
22	0	634	438	835	767	16	0	750	743	725	635	715	683	685	685	491
22	1	1828	1820	634	438	16	1	393	271	510	540	865	811	1177	1115	491
22	2	750	750	1828	1820	16	2	341	275	254	254	461	434	434	434	491
22	3	714	474	750	750	16	3	341	275	254	254	461	434	434	434	491
22	4	410	351	714	474	16	4	604	442	1213	1245	461	434	434	434	491
22	5	547	523	410	351	16	5	604	442	1213	1245	461	434	434	434	491
22	6	1029	1022	547	523	17	0	1066	1636	472	472	1309	1241	1018	1018	491
22	7	708	646	1029	1022	17	1	933	542	472	472	1309	1241	1018	1018	491
24	0	1244	1154	708	646	17	2	1057	1085	235	140	1205	1241	1018	1018	491
24	1	333	304	1244	1154	17	3	1273	1168	235	140	1205	1241	1018	1018	491
24	2	1839	1830	333	304	17	4	597	510	388	205	390	320	615	606	491
24	3	606	472	1839	1830	17	5	1005	1005	388	205	390	320	615	606	491
26	0	1065	944	606	472	17	6	1203	1005	310	226	524	584	578	630	491
26	1	1230	1154	1065	944	17	7	657	688	682	651	521	885	885	630	491
26	2	667	659	1230	1154	18	7	1012	1028	715	686	645	597	284	315	491
26	3	1290	1406	667	659	18	8	716	671	512	467	271	432	1900	1856	491
26	4	659	659	1290	1406	18	9	303	455	1459	1449	276	263	776	832	491
K	H	FOBS	FCALC	FOBS	FCALC	L=		555	513	1165	1173	723	642	552	625	426
0	0	1023	1715	1023	1715			393	393	1074	1075	360	378	278	275	411
								757	665	484	484	561	750	571	275	758
										441	484	374	412	416	452	426

0	2	554	502	15	3	612	535	14	0	540	606	11	7	336	415	12	5	782	834	4	2	1015	1088
C	4	856	732	15	5	451	408	14	1	855	797	13	0	221	251	12	2	598	682	4	2	693	576
0	6	328	339	15	4	520	445	14	2	633	834	12	2	422	323	12	8	357	293	4	4	754	874
0	10	321	417	15	6	804	721	14	4	633	651	12	3	270	217	13	0	1335	1158	4	4	754	874
1	0	1030	1025	15	7	722	682	14	5	760	756	12	6	223	425	13	0	870	842	4	4	867	958
1	1	1106	1175	15	8	303	267	14	6	332	392	13	7	425	360	13	3	372	287	4	4	604	711
1	2	453	556	20	1	305	457	14	7	356	307	13	8	330	163	13	4	542	955	4	4	441	441
1	3	1627	1791	20	3	258	272	15	0	595	505	14	1	253	325	13	5	272	35	4	4	441	441
1	5	726	756	20	5	375	434	15	2	567	603	14	2	392	280	13	6	242	355	4	4	441	441
1	6	254	147	20	7	338	108	15	4	1903	1890	14	3	392	856	13	6	242	355	4	4	441	441
1	7	1149	1255	21	0	338	108	15	2	288	1890	14	2	392	856	13	6	242	355	4	4	441	441
1	11	801	837	21	2	506	511	15	3	1310	1279	14	3	442	473	14	0	1341	1392	4	4	533	547
2	0	485	457	21	3	765	677	15	4	750	807	15	7	605	565	14	1	360	337	4	4	533	547
2	1	452	438	21	4	463	471	15	5	1131	1135	15	6	605	565	14	2	304	189	4	4	533	547
2	3	216	225	21	6	761	666	15	9	514	501	15	5	248	26	14	3	385	443	4	4	533	547
2	5	059	719	21	7	828	814	16	1	618	522	16	0	897	545	14	4	900	576	4	4	533	547
2	6	413	384	23	1	356	370	16	4	556	1025	16	1	326	326	15	7	562	569	4	4	533	547
2	9	447	240	23	2	428	507	16	5	310	724	16	2	248	326	15	4	1241	1627	4	4	533	547
2	11	275	87	23	2	205	446	16	5	310	257	16	2	854	508	15	2	285	1627	4	4	533	547
3	0	990	1188	23	5	501	574	17	0	857	849	16	4	854	508	15	4	746	737	4	4	533	547
3	1	263	265	23	5	1260	1266	17	0	250	163	16	5	518	550	16	0	1179	1145	4	4	533	547
3	2	390	308	27	5	1260	1266	17	1	1543	1542	16	6	1400	1467	16	0	626	673	4	4	533	547
3	3	201	180	27	5	1260	1266	17	4	362	452	17	8	534	581	16	4	490	505	4	4	533	547
3	3	412	455	27	5	1260	1266	17	5	360	218	17	1	444	440	16	4	825	505	4	4	533	547
3	10	456	330	27	5	1260	1266	17	7	645	554	17	2	278	319	17	2	458	587	4	4	533	547
3	11	252	348	27	5	1260	1266	17	8	374	379	17	3	245	545	17	3	295	71	4	4	533	547
4	0	1139	1188	27	5	1260	1266	18	0	455	557	17	4	245	47	17	5	454	415	4	4	533	547
4	1	320	456	27	5	1260	1266	18	2	414	295	18	0	773	438	17	6	627	515	4	4	533	547
4	2	647	341	27	5	1260	1266	18	4	246	120	18	1	850	712	18	0	627	686	4	4	533	547
4	3	647	656	27	5	1260	1266	18	6	688	135	18	2	635	676	18	2	568	377	4	4	533	547
4	4	647	656	27	5	1260	1266	18	7	333	20	18	3	645	854	18	3	400	576	4	4	533	547
4	5	351	280	27	5	1154	1217	15	0	404	561	16	4	336	491	18	5	273	103	4	4	271	346

L = 2